## Review: Absolute Dating

1. What does a radioactive element release/give off? $\qquad$
2. A half-life is the $\qquad$ it takes for $\qquad$ of the radioactive elements to $\qquad$ .
3. If a rock contains $\mathbf{5 0 \%}$ radioactive material, how many half-lives have gone by?
4. If a rock contains $\mathbf{2 5 \%}$ radioactive material, how many half-lives have gone by?
5. If a rock contains $\mathbf{1 2 . 5} \%$ radioactive material, how many half-lives have gone by?
6. A radioactive element has a half-life of $\mathbf{1 5}$ years. A rock contains $\mathbf{2 5 \%}$ radioactive material.

First figure out how many half-lives have passed if $25 \%$ radioactivity remains $\qquad$
Then multiply the length of the half-life by the number that has passed to give you the age of the rock. How old is the rock?
7. A radioactive element has a half-life of 20 years. A rock contains $\mathbf{6 . 2 5} \%$ radioactive material. How old is the rock?


## - Use the Decay curve for Tritium.

8. As time increases, to the amount of radioactivity in Tritium
$\qquad$
9. How old is a rock that has $50 \%$ Tritium? $\qquad$
10. How old is a rock that has $25 \%$ Tritium? $\qquad$
11. How many years are in the half-life of Tritium? $\qquad$


## - Use the Carbon 14 Decay curve

12. How old is a rock with $25 \%$ Carbon 14 $\qquad$
13. How old is a rock with 50 \% Carbon 14 $\qquad$
14. Estimate the number of years in the half-life of Carbon 14 using the graph $\qquad$ .
15. Carbon 14 is used to find the age of organic fossils, which means that they were once $\qquad$

Use the table below to help you answer the following questions.

| Original Radioactive material | Daughter Material | Half-life |
| :--- | :--- | :--- |
| $\mathrm{U}-235$ | $\mathrm{~Pb}-207$ | 713 Million years |
| $\mathrm{U}-238$ | $\mathrm{~Pb}-206$ | 4.5 billion years |
| $\mathrm{K}-40$ | $\mathrm{Ar}-40$ | 1.3 billion years |

16. A rock contains 25 units of $\mathbf{U}-\mathbf{2 3 5}$ and 75 units of $\mathbf{P b - 2 0 7}$. How old is the rock?
17. A rock contains 50 units of $\mathbf{U}-\mathbf{2 3 8}$ and 50 units of $\mathbf{P b - 2 0 6}$. How old is the rock?
18. A rock contains 12.5 units of $\mathbf{K}-\mathbf{4 0}$ and 87.5 units of Ar-40. How old is the rock?
