Name _____ period _____ DO NOW

Recognize that chemical formulas are used to identify substances and determine the number of atoms of each element in chemical formulas containing subscripts.

Iormulas containing subscripts. Use the chemical reaction below for each of the next questions.

$Ba(NO_3)_2$ + 2NaCl \rightarrow $BaCl_2$ + 2NaNO_3

Date

The reaction above shows what happens when barium nitrate and sodium chloride are mixed. According to the chemical reaction, how many atoms of oxygen are found on the product side of the equation?

- 1. Underline the question.
- 2. Box the important vocabulary words.
- 3. Tell something else you know about this topic.

4. Answer the question in the space below.

Date

In the reaction of barium nitrate and sodium chloride shown how many molecules of sodium chloride are found on the reactant side of the equation?

- 1. Underline the question.
- 2. Box the important vocabulary words.
- 3. Tell something else you know about this topic.

4. Answer the question in the space below.

Date

When barium nitrate and sodium chloride are mixed together they produce barium chlorite and sodium nitrite. What is the total number of atoms in the chemical formula of barium chlorite (BaCl₂)?

1. Underline the question.

- 2. Box the important vocabulary words.
- 3. Tell something else you know about this topic.
- 4. Answer the question in the space below.

Date

Some chemical formulas contain subscripts. The subscripts tell us how many atoms of each element are found in the molecule. How many atoms are in each element of the chemical formula $C_8H_9NO_2$?

- 1. Underline the question.
- 2. Box the important vocabulary words.
- 3. Tell something else you know about this topic.

4. Answer the question in the space below.

Date

Many chemical formulas have coefficients in front of them. This tells us the number of molecules of that compound we have. If you have a coefficient in front of the formula you must multiply the number of atoms in the formula times the coefficient to get the total number of atoms. How many atoms are in each element of the chemical formula $2Fe_2O_3$?

- 1. Underline the question.
- 2. Box the important vocabulary words.
- 3. Tell something else you know about this topic.
- 4. Answer the question in the space below

Q N B M I L K Y W A Y U N I Y E v v TBNW C S RIEP WDAVAIDR ZGJEKOVO ZUODNHU CEMNYYBIY Z HOHUEART ΙΤΡΤ Н ZHUSIVLA Z ESS YENEBULALARW 0 M Y Y YRRCSEQPLZENRONP ΗΑЈΡΗΧΧΕΗΒΙΑΟΚΗΟ UDWDPAGTDIOOJZXHXNY G P D X T M Z S K Y A S U K W C Y N E O N Y Y N I I M O P R S UPPNRRVROHYBXIND A R S F N B C C E S Q Q T K I T E EEEPNOID ΥΥΈ ΗΑΑΙυΗΥ RMRIJLJAGUPJIU SZAT IDSATURNKL K M P N O F P X N C J R A V S E D L F N Y A J Y H W E X U K C A M A I A Jupiter Mercury Earth Neptune Venus Comet Saturn Uranus Meteor Mars Pluto Galaxy Nebula Universe Asteroids Supernova Blackhole Constellation Orbit Milkyway